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1940 DUKE STREET ALEXANDRIA, VA 22314			SALVITTI, MICHAEL A	
ALEAANDRIA, VA 22314			ART UNIT	PAPER NUMBER
		1796		
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			05/29/2009	ELECTRONIC

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Notice of the Office communication was sent electronically on above-indicated "Notification Date" to the following e-mail address(es):

patentdocket@oblon.com oblonpat@oblon.com jgardner@oblon.com

		Applic	ation No.	Applicant(s)			
Office Action Summary		10/561	,266	CASALINI ET AL	CASALINI ET AL.		
		Exami	ner	Art Unit			
		MICHA	EL A. SALVITTI	1796			
The MA Period for Reply	ILING DATE of this commu	nication appears on	the cover sheet wit	h the correspondence a	ddress		
A SHORTENE WHICHEVER I - Extensions of time after SIX (6) MON - If NO period for rej - Failure to reply wit Any reply received	D STATUTORY PERIOD F S LONGER, FROM THE M may be available under the provision THS from the mailing date of this com ply is specified above, the maximum s hin the set or extended period for repl by the Office later than three months an adjustment. See 37 CFR 1.704(b).	MAILING DATE OF s of 37 CFR 1.136(a). In no munication. tatutory period will apply an y will, by statute, cause the	THIS COMMUNIC be event, however, may a red d will expire SIX (6) MONT application to become ABA	CATION. The ply be timely filed THS from the mailing date of this of the capacity of the cap			
Status							
2a)⊠ This action 3)□ Since this	ive to communication(s) filon is FINAL . s application is in condition accordance with the pract	2b)∏ This action in for allowance exce	s non-final. ept for formal matte	· · · · ·	e merits is		
Disposition of Cla	iims						
4a) Of the 5) ☐ Claim(s) 6) ☑ Claim(s) 7) ☐ Claim(s)	1-16 is/are pending in the above claim(s) is/a is/are allowed. 1-16 is/are rejected. is/are objected to. are subject to restricts	are withdrawn from					
10)☐ The draw Applicant	ification is objected to by the ing(s) filed on is/are may not request that any objections.	: a) accepted or ection to the drawing(s) be held in abeyand	ce. See 37 CFR 1.85(a).			
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d). 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.							
Priority under 35	-	•					
12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f). a) ☐ All b) ☐ Some * c) ☐ None of: 1. ☐ Certified copies of the priority documents have been received. 2. ☐ Certified copies of the priority documents have been received in Application No 3. ☒ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)). * See the attached detailed Office action for a list of the certified copies not received.							
2) D Notice of Draftsp	nces Cited (PTO-892) erson's Patent Drawing Review (osure Statement(s) (PTO/SB/08) Date	PTO-948)	Paper No(s)	ummary (PTO-413))/Mail Date formal Patent Application _·			

DETAILED ACTION

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-3 and 7 rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. Patent No. 4,493,922 to *Echte et al.*, with supporting evidence provided by U.S. Patent No. 5,039,714 to *Kasahara et al.*

Regarding claim 1: Echte teaches rubber-reinforced (polybutadinene or butadiene/styrene (col. 1, lines 10-15) vinyl aromatic (polystyene; col. 1, line 39) copolymers. These polymers have strictly bimodal morphology, as demonstrated by (b1) particles (equivalent to "capsule" or "core-shell" particles of instant application; col. 1, lines 47-48) and (b2) particles (equivalent to "salami" particles of instant application; col. 1, lines 54-55). The (b2) particles are known in the art as "salami particles (see *Kasahara* col. 2, lines 55-56 for supporting evidence).

Echte teaches the rigid polymeric matrix (polystyrene) from 70-97% by weight (col. 1, lines 41-43). The rubbery phase dispersed inside of the rigid polymeric matrix comprisies 5-40% by weight (col. 1, line 49). The particles with a capsule or "core-shell" morphology constitute 60-95% by weight (col. 1, lines 47-48 and col. 2, lines 17-18), and particles with salami morphology comprise 5-40% by weight (col. 1, lines 49-50). In

the case where the claimed ranges "overlap or lie inside ranges disclosed by the prior art" a *prima facie* case of obviousness exists. *In re Wertheim*, 541 F.2d 257, 191USPQ 90 (CCPA 1976). See MPEP § 2144.05.

Echte is silent regarding the difference between the solubility parameter according to Hildebrand of the elastomer as being higher than or equal to 0.5. 40/60 Polystyrene-polybutadiene block is admitted in the instant specification as having a known Hildebrand parameter of δ = 8.7, which is higher than 0.5 (page 12, lines 5-12 of instant specification). Echte uses 60-95% polybutadiene (with styrene comprising the balance) as capsule particles (col. 1, lines 47-48 and col. 2, lines 17-25), suggesting that a capsule particle with a high Hildebrand value is present. Echte further teaches the "salami" particle equivalent as having 5% to 40% polybutadiene (col. 2, lines 27-30). Styrene and butadiene have different Hildebrand parameters, and changing the ratios this drastically would intrinsically create a difference in the Hildebrand parameters larger than 0.5. Unless it can be shown otherwise, the Office assumes that the properties disclosed are inherent to the composition. "The discovery of a previously unappreciated property of a prior art composition, or of a scientific explanation for the prior art's functioning, does not render the old composition patentably new to the discoverer." Atlas Powder Co. v. Ireco Inc., 190 F.3d 1342, 1347, 51 USPQ2d 1943, 1947 (Fed. Cir. 1999). See MPEP § 2112.

Regarding claim 2: Echte teaches core-shell particles with an average diameter ranging from 0.2-0.6 μ m and salami structured particles with an average diameter ranging from 2-8 μ m. In the case where the claimed ranges "overlap or lie inside

ranges disclosed by the prior art" a *prima facie* case of obviousness exists. *In re Wertheim*, 541 F.2d 257, 191USPQ 90 (CCPA 1976). See MPEP § 2144.05.

Regarding claim 3: Echte teaches 60-95% core-shell polymers (b1), based on the weight of 1,3 alkadiene monomer (col. 1, lines 47-49). These core-shell compounds may contain styrene (col. 2, line 20) as a mono-ethylenically unsaturated monomer, with styrene present from 5% to 40%, calculated as the balance of the mass of the particle. In the case where the claimed ranges "overlap or lie inside ranges disclosed by the prior art" a prima facie case of obviousness exists. In re Wertheim, 541 F.2d 257, 191USPQ 90 (CCPA 1976). See MPEP § 2144.05.

Regarding claim 7: Echte teaches elastomeric products capable of providing a rubbery phase dispersed in the rigid polymeric matrix (col. 1, lines 24-55), in the form of grafted and occluded particles with a "salami" morphology (see *Kasahara* col. 2, lines 55-56 for definition), comprising homopolymers and/or copolymers of 1,3 alkadienes (polybutadiene; col. 1, lines 45-50) incompatible with the elastomeric products which provide the capsule rubbery phase. Unless it can be shown otherwise, the Office assumes that the incompatibility of the salami and elastomeric products are inherent to the composition taught by *Echte*.

Claims 4-6 and 8 are rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. Patent No. 4,493,922 to *Echte et al* in view of U.S. Patent No. 6,545,090 to *Demirors et al*.

Regarding claim 4: Echte teaches the styrene-butadiene (S-B) copolymers (col. 2, line 20) according to claim 3.

Echte is silent regarding the molecular weights of S-B type polymers. *Demirors* teaches elastomeric products (rubbers) having a polystyrene block having an average molecular weight between 60,000-80,000 (col. 5, lines 8-17), and a polybutadiene having an average molecular weight between 100,000-1,000,000 (col. 4, lines 1-10). These references are analogous art in that they are drawn to the same field of endeavor, namely the synthesis of impact-resistant polystyrenes, via blends of rubbery particles added to a polystyrene matrix. At the time of the invention, it would have been obvious to a person having ordinary skill in the art to incorporate elastomers with molecular weights in these ranges, as taught by *Demirors* into the composition of *Echte*, with the motivation of optimizing the Mooney viscosities of the resultant polymers (*Demirors*, col. 4, lines 11-24). This would enable a person having ordinary skill in the art to optimize the processing parameters, resulting in an improved product.

Regarding claim 5: Echte teaches the styrene-butadiene (S-B) copolymers (col. 2, line 20).

Echte is silent regarding the ratio of styrene to butadiene present in the S-B type polymer. Demirors teaches a styrene block with 20-80% by weight of the copolymer (col. 5, lines 18-24). At the time of the invention, it would have been obvious to a person having ordinary skill in the art incorporate 10-50% by weight styrene block copolymers into the invention of Echte, with the motivation of obtaining the proper proportions of small and large rubber particles (Demirors, col. 5, lines 18-20), which is

stated to result in a good combination of impact resistance, tensile strength and surface gloss (col. 6, lines 5-30).

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Regarding claim 6: Echte teaches the styrene-butadiene (S-B) copolymers (col. 2, line 20).

Echte is silent regarding the styrene content equal to 40% by weight and a viscosity in solution ranging from 30-50 cPs. *Demirors* teaches a styrene-butadiene block copolymer having a styrene content ranging from 20-80% (col. 5, lines 18-25), having a viscosity measured at 40-400 cps (col. 5, lines 25-35). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to prepare the polymer composition of *Echte* with the parameters taught by *Demirors*, with the motivation of obtaining a high-impact polystyrene composition which is sufficiently solid to be handled and processed in a normal fashion (col. 5, lines 40-42).

Regarding claim 8: Echte teaches the composition of claim 7.

Echte is silent regarding the elastomeric product as polyisoprene, with a viscosity of 100-1000 cPs. *Demirors* teaches isoprenes as a preferred 1,3 conjugated diene (col. 3 line 65 through col. 4, line 10). Viscosities below 300 centipoise are taught (col. 4, lines 12-24). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to use a polyisoprene elastomer with a viscosity ranging from 100-1000 cps, with the motivation of obtaining a high-impact polystyrene composition which is sufficiently solid to be handled and processed in a normal fashion (col. 5, lines 40-42).

Claims 9-16 are rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. Patent No. 4,493,922 to *Echte et al.* in view of U.S. Patent No. 6,545,090 to *Demirors et al.*

Regarding claim 9: Echte teaches a process for the preparation of rubber-reinforced vinyl aromatic copolymer with a bimodal morphology, consisting of a rigid polymeric matrix (polystyrene) from 70-97% by weight (col. 1, lines 41-43). The rubbery phase dispersed inside of the rigid polymeric matrix comprisies 5-40% by weight (col. 1, line 49). The particles with a capsule or "core-shell" morphology constitute 60-95% by weight (col. 1, lines 47-48 and col. 2, lines 17-18), and particles with salami morphology comprise 5-40% by weight (col. 1, lines 49-50). In the case where the claimed ranges "overlap or lie inside ranges disclosed by the prior art" a prima facie case of obviousness exists. In re Wertheim, 541 F.2d 257, 191USPQ 90 (CCPA 1976). See MPEP § 2144.05.

Echte is silent regarding a process comprising the method of preparation, as set forth by steps a-c of the instant claim. *Demirors* teaches a process wherein 5-10% by weight of rubber is dissolved (col. 9, lines 5-6) This rubber is preferably 1,3 conjugated diene homopolymer (i.e. 100% 1,3 alkadiene monomer, 0% mono-ethylenically unsaturated monomer; col. 3, lines 65 through col. 4, line 11). The resulting solution is polymerized at a temperature ranging from 60-190°C in the presence of chain transfer agents (*Demirors*, col. 10, lines 6-17). Recovery of the polymer is taught (col. 9, lines 55-61). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to prepare the polymers taught by *Echte* via the continuous

method of *Demirors*, with the motivation of improving the mechanical properties of the product, in particular product impact resistance (*Demirors*, col. 9, lines 25-29).

Neither reference discloses the solubility parameters of the components, despite using 1,3 alkadienes and styrene components, as taught by the instant claim. The Hildebrand solubility parameters δ_1 and δ_2 are intrinsic to the capsule and salami particles, respectively, and will be larger than 0.5 based on the ratios taught by *Echte*. Thus, unless it can be shown otherwise, the Office assumes that the properties disclosed are inherent to the composition, as Hildebrand solubility parameter is a property intrinsic to the component. "The discovery of a previously unappreciated property of a prior art composition, or of a scientific explanation for the prior art's functioning, does not render the old composition patentably new to the discoverer." *Atlas Powder Co. v. Ireco Inc.*, 190 F.3d 1342, 1347, 51 USPQ2d 1943, 1947 (Fed. Cir. 1999). See MPEP § 2112.

Regarding claim 10: Echte teaches a process for the preparation of rubber-reinforced vinyl aromatic copolymer with a bimodal morphology, consisting of a rigid polymeric matrix (polystyrene) from 70-97% by weight (col. 1, lines 41-43). The rubbery phase dispersed inside of the rigid polymeric matrix comprisies 5-40% by weight (col. 1, line 49). The particles with a capsule or "core-shell" morphology constitute 60-95% by weight (col. 1, lines 47-48 and col. 2, lines 17-18), and particles with salami morphology comprise 5-40% by weight (col. 1, lines 49-50). In the case where the claimed ranges "overlap or lie inside ranges disclosed by the prior art" a *prima facie* case of

obviousness exists. *In re Wertheim*, 541 F.2d 257, 191USPQ 90 (CCPA 1976). See MPEP § 2144.05.

Demirors teaches a process wherein 5-10% by weight of rubber is dissolved (col. 9, lines 5-6) This rubber is preferably 1,3 conjugated diene homopolymer (i.e. 100% 1,3 alkadiene monomer, 0% mono-ethylenically unsaturated monomer; col. 3, lines 65 through col. 4, line 11). The resulting solution is polymerized at a temperature ranging from 60-190°C in the presence of chain transfer agents (*Demirors*, col. 10, lines 6-17). Recovery of the polymer is taught (col. 9, lines 55-61). Prepolymerization conditions are maintained such that phase inversion occurs (*Demirors*, col. 9, lines 30-35). The reaction may contain suspending agents such as butyl stearate (col. 9, lines 60-68). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to prepare the polymers taught by *Echte* via the continuous method of *Demirors*, with the motivation of improving the mechanical properties of the product, in particular product impact resistance (*Demirors*, col. 9, lines 25-29).

Neither reference discloses the solubility parameters of the components, despite using 1,3 alkadienes and styrene components, as taught by the instant claim. The Hildebrand solubility parameters δ_1 and δ_2 are intrinsic to the capsule and salami particles, respectively, and will be larger than 0.5 based on the ratios taught by *Echte*. Thus, unless it can be shown otherwise, the Office assumes that the properties disclosed are inherent to the composition, as Hildebrand solubility parameter is a property intrinsic to the component. "The discovery of a previously unappreciated property of a prior art composition, or of a scientific explanation for the prior art's

functioning, does not render the old composition patentably new to the discoverer." *Atlas Powder Co. v. Ireco Inc.*, 190 F.3d 1342, 1347, 51 USPQ2d 1943, 1947 (Fed. Cir. 1999). See MPEP § 2112.

Regarding claim 11: Echte teaches styrene as a vinyl aromatic monomer, but is silent regarding the method of claim 9.

Echte teaches styrenyl derivatives having the formula given in claim 11, wherein n=0 and R is hydrogen (styrene). Demirors teaches the method of claim 9, and discloses alkyl and halogenated styrenyl derivatives of the invention (col. 3, lines 15-54), thereby teaching a finite number of identified predictable solutions, with a reasonable expectation of success (See MPEP § 2141 "E"). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to prepare the polymers taught by Echte via the continuous method of Demirors, with the motivation of improving the mechanical properties of the product, in particular product impact resistance (Demirors, col. 9, lines 25-29).

Regarding claim 12: Echte teaches a process for making a composition having the same components as claim 9.

Echte is silent regarding rubbers dissolved in the monomers, possibly in the presence of an inert solvent in quantities ranging from 5-20% by weight with respect to the total. *Demirors* teaches 5-15% of rubber dissolved in monomers (col. 6, lines 55-60), possibly in the presence of an inert solvent (col. 7, lines 5-28). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to prepare the polymers taught by *Echte* via the continuous method of *Demirors*, with the

motivation of improving the mechanical properties of the product, in particular product impact resistance (*Demirors*, col. 9, lines 25-29).

Regarding claim 13: Echte teaches a process for making a composition having the same components as claim 9.

Echte is silent regarding the temperature of dissolution. Demirors teaches dissolution without disclosing heating (col. 6, lines 54-60), suggesting that the dissolution temperature may be room temperature, which is below 100°C. Alternatively, the polymerization may occur as low as 60°C, suggesting that the dissolution temperature was below the initiaton temperature. At the time of the invention, it would have been obvious to a person having ordinary skill in the art to prepare the polymers taught by Echte via the continuous method of Demirors, with the motivation of improving the mechanical properties of the product, in particular product impact resistance (Demirors, col. 9, lines 25-29).

Regarding claim 14: Echte teaches a process for making a composition having the same components as claim 9.

Echte is silent regarding the temperature and pressure of the reaction. *Demirors* teaches polymerizations occurring between 60 and 190°C (col. 10, line 16). These polymerizations are suspension polymerization techniques (col. 5, lines 25-30). The reaction occurs in a stirred-tank reactor (col. 9, line 39). *Demirors* does not specify the pressure, suggesting the vessel is not pressurized, and has a pressure of 1.01 bar (atmospheric pressure). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to prepare the polymers taught by *Echte* via the

continuous method of *Demirors*, with the motivation of improving the mechanical properties of the product, in particular product impact resistance (*Demirors*, col. 9, lines 25-29).

Regarding claim 15: Echte teaches a process for making a composition having the same components as claim 9.

Echte is silent regarding the quantities of initiators added. Demirors teaches use of initiators at 100-1,500 ppm (0.01 to 0.15% by weight; col. 8, lines 37-47). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to prepare the polymers taught by Echte via the continuous method of Demirors, with the motivation of improving the mechanical properties of the product, in particular product impact resistance (Demirors, col. 9, lines 25-29).

Regarding claim 16: Echte teaches a process for making a composition having the same components as claim 9.

Echte is silent regarding the quantities of chain transfer agents added. *Demirors* teaches 0.0001-0.5% by weight of chain transfer agents (col. 10, lines 6-17). At the time of the invention, it would have been obvious to a person having ordinary skill in the art to prepare the polymers taught by *Echte* via the continuous method of *Demirors*, with the motivation of improving the mechanical properties of the product, in particular product impact resistance (*Demirors*, col. 9, lines 25-29).

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Response to Arguments

A) A typographical change to claim 10 has been noted; no new matter has been added by this change, and the objection to Claim 10 has been removed.

B) Applicant's arguments (See page 8 of "Remarks") filed 02/27/2009 have been fully considered but they are not persuasive. Regarding the rejection of claims 1-3 and 7 under U.S.C. 103(a), the applicant states that *Echte* is silent regarding the method of production, in accordance with a method set forth by the instant invention. However, the product of *Echte* is directed towards the claimed invention (claims 1-3 and 7), and not the process. These product-by-process limitations are not included in claims 1-3 and 7.

The applicant further states that the Hildebrand solubility parameter is not taught by *Echte*. While this is acknowledged, unless it can be otherwise shown, the Hildebrand parameters are properties inherent to the components of the composition. As such, the measurement of this value or the recognition of a previously unrecognized property does not constitute patentability, as the composition of instant claims 1-3 and 7 contains overlapping ranges for the components of *Echte*'s composition.

C) The applicant argues (page 8, bottom of page) that butyl rubber is not polybutadiene. It is noted that polybutadiene is not butyl rubber in view of the definition provided by *Hawley. Echte* teaches polybutadiene which is a 1,3 alkadiene as required by claims 3 and 7. In response to applicant's argument that the references fail to show certain features of applicant's invention, it is noted that the features upon which applicant relies (i.e., rubber is claimed, not butyl rubber) are not recited in the rejected

claim(s). Although the claims are interpreted in light of the specification, limitations from the specification are not read into the claims. See *In re Van Geuns*, 988 F.2d 1181, 26 USPQ2d 1057 (Fed. Cir. 1993).

- D) The applicant argues (page 9, top half of page) that *Kasahara* does not disclose the Hildebrand parameter. As stated above, this parameter is inherent to the type of polymers used in the composition. *Kasahara* was cited as an evidentiary reference to show equivalence of the terms salami particles and "cell/coil particles".
- E) The applicant argues (page 9, bottom half of page) that while *Demirors* demonstrates overlapping ranges for weight percentages and properties not cure the deficiency of the missing Hildebrand relationship. As stated above, the Hildebrand relationship is inherent to the polymer composition used.
- F) The applicant argues (pages 10-11) that the Hildebrand relationship of claim 1 is not taught in *Echte* or *Demirors*. Thus, unless it can be shown otherwise, the Office assumes that the properties disclosed are inherent to the composition. "The discovery of a previously unappreciated property of a prior art composition, or of a scientific explanation for the prior art's functioning, does not render the old composition patentably new to the discoverer." *Atlas Powder Co. v. Ireco Inc.*, 190 F.3d 1342, 1347, 51 USPQ2d 1943, 1947 (Fed. Cir. 1999). See MPEP § 2112.

Conclusion

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

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A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event, however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Correspondence

Any inquiry concerning this communication or earlier communications from the examiner should be directed to MICHAEL A. SALVITTI whose telephone number is (571)270-7341. The examiner can normally be reached on Monday-Thursday 8AM-7PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Mark Eashoo can be reached on (571) 272-1197. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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/Mark Eashoo/ /M. A. S./

Supervisory Patent Examiner, Art Unit 1796 Examiner, Art Unit 1796